

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of

SEGAL, David Leslie

Filed: (Herewith)

For: PLASMA ASSISTED PROCESSING OF GAS

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PRELIMINARY AMENDMENT  
for  
CONTINUATION APPLICATION

To the Assistant Commissioner for Patents  
Washington, D.C. 20231

Sir:

Submitted herewith as a new continuation application is a duplicate copy of the parent application, Serial No. 09/582,986, filed July 10, 2000.

Please amend, as follows:

IN THE SPECIFICATION:

Page 1, after the Title, insert --This is a continuation of Application Serial No. 09/582,986, filed July 10, 2000.--

IN THE CLAIMS:

Cancel claims 1-17 and substitute the following claims in lieu thereof:

18.(New) A method for the simultaneous removal of nitrogen oxides and carbonaceous combustion products from gases, said method comprising the steps of:

(a) feeding gases containing nitrogen oxides and carbonaceous combustion products into a reaction chamber which contains a gas

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permeable bed of an active material which includes a mixed metal oxide material having the general formula  $A_{2-x}A^1_xB_{1-y}B^1_yO_4$ ;

(b) exciting said gases into a plasma state for creating plasma activated species;

(c) moving said gases and activated species through said gas permeable bed of active material; and

(d) exposing said metal oxide material to said gases and activated species for creating catalytic action enhanced by said plasma activated species and causing interaction between and simultaneous removal of said nitrogen oxides and carbonaceous products from said gases.

19. (New) The method as defined in claim 18 including the additional step of:

(a) aligning said reactor chamber in-line with the exhaust system of an internal combustion engine for receiving exhaust gases therefrom.

20. (New) The method as defined in claim 18 including the step of selecting the mixed metal oxide active material from the group consisting of  $La_{1.8}Ba_{0.2}CuO_4$ ;  $La_{1.7}Sr_{0.3}Cu_{0.9}V_{0.1}O_4$ ;

$La_{1.9}K_{0.1}Cu_{0.7}Cr_{0.3}O_4$ ;  $La_{1.8}Ba_{0.2}Cr_{0.7}V_{0.3}O_4$  and  $La_{1.9}K_{0.1}Cu_{0.95}V_{0.05}O_4$ .

21. (New) The method as defined in claim 20 including the step of making said metal oxide material in the form of an agglomeration of bodies in the form of spheres, regularly or irregularly shaped pellets or hollow extrudates.

22. (New) The method as defined in claim 21 including the

step of exciting said gases into a plasma state separate from and preceding the bed of mixed metal oxide active material.

23. A plasma assisted reactor for the simultaneous removal of nitrogen oxides and carbonaceous combustion products from exhaust gases, comprising a reactor chamber adapted to be connected into a gas exhaust system, a gas permeable bed of an active material contained within the reactor chamber, conduit for causing exhaust gases to pass through the bed of active material, and means for exciting into a plasma state exhaust gases passing through the bed of active material for creating plasma activated species and causing interaction between and simultaneous removal of said nitrogen oxides and carbonaceous combustion products, the bed of active material including a mixed metal oxide material having the general formula  $A_{2-x}A^1_xB_{1-y}B^1_yO_4$ .

24. (New) A reactor according to claim 23, wherein the components A  $A^1$  of the mixed metal oxide material are selected from the group consisting of elements comprising La, Sr, Ba and K and the components B  $B^1$  of the mixed metal oxide material are selected from the group of elements comprising Co, Mn, Cr, Cu, Mg and V.

25. A reactor according to claim 24, wherein the mixed metal oxide active material is selected from the group consisting of  $La_{1.8}Ba_{0.2}CuO_4$ ;  $La_{1.7}Sr_{0.3}Cu_{0.9}V_{0.1}O_4$ ;  $La_{1.9}K_{0.1}Cu_{0.7}Cr_{0.3}O_4$ ;  $La_{1.8}Ba_{0.2}Cr_{0.7}V_{0.3}O_4$  and  $La_{1.9}K_{0.1}Cu_{0.95}V_{0.05}O_4$ .

26. A reactor according to claim 25, wherein said means for exciting the gases to the plasma state comprises at least two electrodes in contact with the bed of active material and means for applying to the electrodes a potential difference sufficient to excite the exhaust gases to a plasma state in the interstices of the bed of active material.

27. Apparatus for the simultaneous removal of nitrogen oxides and carbonaceous combustion products from gases, said apparatus comprising a reaction chamber, means for feeding gases containing nitrogen oxides and carbonaceous material into said reaction chamber, said reaction chamber containing a gas permeable bed of an active material which includes a mixed metal oxide material having the general formula of  $A_{2-x}A^1_xB_{1-y}B^1_yO_4$ ; means for exciting said gases into a plasma state for creating plasma activated species through said gas permeable bed of active material; and means for exposing said mixed metal oxide material to said gases and activated species for creating catalytic action enhanced by said plasma activated species and causing interaction between and simultaneous removal of said nitrogen oxides and carbonaceous products from said gases.

28. (New) A reactor according to claim 27, wherein the mixed metal oxide active material is selected from the group consisting of  $La_{1.8}Ba_{0.2}CuO_4$ ;  $La_{1.7}Sr_{0.3}Cu_{0.9}V_{0.1}O_4$ ;  $La_{1.9}K_{0.1}Cu_{0.7}Cr_{0.3}O_4$ ;  $La_{1.8}Ba_{0.2}Cr_{0.7}V_{0.3}O_4$  and  $La_{1.9}K_{0.1}Cu_{0.95}V_{0.05}O_4$ .

R E M A R K S

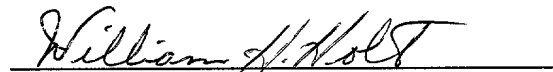
An Abstract is attached hereto.

The specification has been amended to refer to the parent application.

Newly presented claims 18-28 are directed to applicant's methods and apparatus for causing interaction and simultaneous removal of nitrogen oxides and carbonaceous products from gases, particularly exhaust gases from internal combustion engines.

Applicant submits that the methods and apparatus are directed to the same invention and restriction would not be proper. Accordingly, action on the merits of all claims is believed to be in order and is courteously solicited. In the event that the Examiner believes otherwise, the courtesy of a telephone call would be appreciated.

Respectfully submitted,



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